# Summary of Initial Risk Assessment Report

## **Tribromomethane; bromoform** CAS No: 75-25-2

PRTR No of Japan: 222

This substance is assessed based on Guideline for Initial Risk Assessment Version 2.0

#### 1. General Information

## 1.1 Physico-chemical properties

Appearance	Colorless to pale yellow liquid
Melting point	7.5 (degC)
Boiling point	149-150 (degC)
Water solubility	3.1 g/L (25 degC)
Henry's constant	54.2 Pa*m <sup>3</sup> /mol (5.35*10 <sup>-4</sup> atm*m <sup>3</sup> /mol) (25degC, estimated)
Octanol/water partition coefficient (log Kow)	1.19 (measured), 1.18 (estimated)
Soil adsorption coefficient	Koc = 35 (estimated)

#### 1.2 Environmental fate

	1.2 Environmental fate					
Bioaccumulation	Exhibits little to no bioaccumulation Bioconcentration factor (BCF): 7.1-21 (0.1 mg/L), 7.7-19 (0.01 mg/L) (carp), measured					
	Disconcentration factor (DCF): 7.11 21 (O.11 mg/2), 7.17 17 (O.01 mg/2) (early), measures					
	Non-biodegradable					
Biodegradation	Alhough tribromomethane is non-biodegradable in aerobic conditions, it is expected to be biodegraded in specific conditions associated with long-term acclimatization when its concentration is low.					
	(In air)					
	Reaction with OH radical:					
	Reaction rate constant is $4.3*10^{-14}$ cm <sup>3</sup> /molecule-sec. (25 degC, estimated)					
	The half-life is 0.5-1 year, given OH radical concentration of 5*10 <sup>5</sup> -1*10 <sup>6</sup> molecule/cm <sup>3</sup> .					
	Reaction with ozone:					
Stability in the	No data					
environment	Reaction with nitrate radical:					
	Reaction rate constant is 1.3*10 <sup>-17</sup> cm <sup>3</sup> /molecule-sec. (25 degC, measured)					
	The half-life is 0.7-7 years, given nitrate radical concentration of 2.4*10 <sup>8</sup> -2.4*10 <sup>9</sup> molecule/cm <sup>3</sup>					
	(10-100 ppt).					
	(In water)					
	Tribromomethane is not hydrolyzed.					
Environmental fate	When released to the aquatic environments, tribromomethane is expected to be mainly removed from water by volatilization to air. Removal by biodegradation is not considered to be important.					

#### 2. Sources of release to the environment

#### 2.1 Annual production, import, export and domestic supply in 2004 (ton/year)

Production	Import	Export	Domestic supply	Remarks
18				

#### **2.2 Uses**

Tribromomethane is mainly used for geological analysis and dense-medium separation process.

## 2.3 Release from the industries within the scope of PRTR system (in 2003)

Release sources		Air (ton)	Waters (ton)	Soil (ton)	Remarks	
Listed	Reported release	0.88	0	0		
Lis	Release outside notification	0.11	0.36	0	Release to river: 7.5 tons	
	Release outside notification from non listed industry		1.6	0	(All aquatic releases were to rivers.)	
Household	Households		5.5	0		
Mobile sources						
Total		3.3	7.5	0		

#### 2.4 Releases from other sources

No information about the substance is available.

#### 2.5 Main release route

Tribromomethane is expected to be released to the aquatic environments and the air mainly through unintentional production of tribromomethane as a result of chlorine disinfection at water purification plants.

## 3. Exposure Assessment

#### 3.1 Measured environmental concentration

Media	No. of points detected / No. of points measured	No. of samples detected / No. of samples measured	Detection range	95th percentile	Detection limit	Year of investigation, Institution
Air (microg/m³)	0/12	0/63	nd		0.042- 3.2	1980 Ministry of the Environment
River water (microg/L)	0/12	0/60	nd		0.2-26	1976 Ministry of the Environment
Drinking water (microg/L)	1,442/5,468	/19,765	nd-50	6.0	1-18	2003 Japan Water Works Association
Food (microg/g) (fish)	0/4	0/20	nd		0.005- 0.0065	1976 Ministry of the Environment

nd: Not detected

For calculation of the 95th percentile, data less than the detection limit are replaced with a value of one half of the detection limit.

#### 3.2 Estimated environmental concentration

Media	Estimated concentration	Description
Air (microg/m³)	0.020	Calculated by mathematical model / Atmospheric Dispersion Model for Exposure and Risk Assessment ver.1.5 (AIST-ADMER)
River water (microg/L)	0.013	Calculated by mathematical model / Integrated River Model to predict the distribution of chemical concentration (IRM1)

## 3.3 Estimated environmental concentration in water (EEC)

	0.013
EEC(microg/L)	Estimated concentration was used for the risk assessment, since no adequate measured concentration was available <sup>1)</sup> .

#### 3.4 Estimated human intake

Int	take route	Concentration used for estimation of intake	Estimated intake (microg/ person/ day)	Estimated intake (microg/ kg-Bodyweight (BW)/ day)		
uc	Air	0.020 (microg/m <sup>3</sup> )	0.40	8.0*10 <sup>-3</sup>		
Inhalation		Estimated concentration was used for the risk assessment, since measured concentrations are outdated.				
	Drinking	6.0 (microg/L)	12	0.24		
	water	The ninety-fifth percentile of measured concentrations in purified water surveyed in 2003 were us for the risk assessment.				
Oral	Food	0.000027 (microg/g)	3.2*10 <sup>-3</sup>	0.000064		
Or		concentration in seawater was	not available, 1/10 of that in is estimated as a product of	ood were not available. Since measured river water was used as concentrations in the concentration in seawater and a BCF.		
	Subtotal		12	0.24		
Total 1	route		12.4	0.25		

<sup>1)</sup> This substance is assessed based on the Guideline for Initial Risk Assessment Version 2.0. Under Version 2.0, a measured concentration and an estimated concentration (calculated by mathematical model) are compared and the larger of two concentrations is used for risk assessment.

#### 4. Hazard assessment

## 4.1 Effects on organisms in the environment

	Acute or Chronic	Species	Endpoint	Concentration
Algae	No data			-
Crustacea	Acute	Penaueus aztecus	96 hours LC <sub>50</sub>	26.0 (mg/L)
Fish	Fish Chronic		> 28 days NOEC Mortality after hatching	4.8 (mg/L)
Key si	tudy		n variegatus) was chosen for west concentration in the haz	the key study because effects ard assessment.

## 4.2 Human health toxicity

Toxicity	Exposure route	Species	Duration / Dose method	Toxic effects (Key study is underlined)	NOAEL or LOAEL	
	Inhalation					
Repeated dose toxicity	Oral	Rat	1 month (feeding)  Reduced body weight gain, hepatic cell vacuolization, decreased serum Glu, ChE, Co ALP, TG, BUN, and LDH		LOAEL: 680 ppm (equivalent to 56.4 mg/kg/day)	
	Dermal					
Reproductive and developmental toxicity						
Carcinogenicity	Evaluation by IARC : Group 3 (not classifiable as to its carcinogenicity to humans)					
Genotoxicity	Considered to 1	Considered to be genotoxic				

## 5. Risk Assessment

## **5.1 Environmental organisms**

Risk characterization	EEC (microg/L)	NOEC * (mg/L)	MOE (NOEC * /EEC)	Product of uncertainty factors	Conclusion	
	0.013 NOEC: 4.8 370,000 100 No immediate co					
	Product of uncertainty factors (UF):  Extrapolation from laboratory test (10) * Toxicity data on one nutritional stage (10) = 100					

#### Recommendation:

The substance is considered to be of no immediate concern for the moment, and a low priority for further work.

NOEC\* means NOEC, LOEC, EC $_{50}$ , etc.

#### 5.2 Human health

## 5.2.1 Repeated dose toxicity

	Intake		Risk characterization			
Exposure route	(microg/kgBW/d ay)	NOAEL (mg/kgBW/day)	МОЕ	Product of uncertainty factors	Conclusion	
Inhalation	8.0*10 <sup>-3</sup>	No adequate data for the risk assessment is available.	Not calculated	Not calculated		
Oral	0.24	LOAEL: 56.4	240,000	10,000	No immediate concern	

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Total	0.25	56.4 (oral )	230,000	10,000	No immediate concern			
Product of uncertainty factors (UF):								
Interspecies (10) * Intraspecies (10) * Using of LOAEL (10) * Duration of test (10) = 10,000								

#### 5.2.2 Reproductive and developmental toxicity

5.2.3 Carcinogenicity		

#### 5.2.4. Recommendation for Human Health

Although there was no adequate toxicity data available to evaluate exposure via the inhalation route, the MOE calculated using total intake from both routes (inhalation and oral) is larger than the product of uncertainty factors. Thus, the substance is considered to be of no immediate concern for the moment and a low priority for further work.

## 6. Supplement

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